

Tellurite uptake by cells of the facultative phototroph *Rhodobacter capsulatus* is a ΔpH -dependent process

Francesca Borsetti^a, Antonio Toninello^b, Davide Zannoni^{a,*}

^aDepartment of Biology, University of Bologna, 42 Irnerio, 40126 Bologna, Italy

^bDepartment of Biological Chemistry, University of Padova, 3 C. Colombo, 35121 Padova, Italy

Received 4 August 2003; revised 3 October 2003; accepted 6 October 2003

First published online 20 October 2003

Edited by Vladimir Skulachev

Abstract The uptake by light-grown cells of *Rhodobacter capsulatus* of the highly toxic metalloid oxyanion tellurite (TeO_3^{2-}) was examined. We show that tellurite is rapidly taken up by illuminated cells in a process which is inhibited by the protonophore carbonyl cyanide-*p*-trifluoromethoxyphenyl-hydrazone (FCCP) and by the K^+/H^+ exchanger nigericin. Notably, the light-driven membrane potential ($\Delta\psi$) is enhanced by $\text{K}_2\text{TeO}_3 \geq 200 \mu\text{M}$. Further, tellurite uptake is largely insensitive to valinomycin, strongly repressed by the sulfhydryl reagent *N*-ethylethylmaleimide (NEM) and competitively inhibited by phosphate. We conclude that tellurite is transported into cells by a ΔpH -dependent, non-electrogenic process which is likely to involve the phosphate transporter (PiT family).

© 2003 Federation of European Biochemical Societies. Published by Elsevier B.V. All rights reserved.

Key words: Membrane potential; Metalloid oxyanion; Protonophore; Tellurite uptake; Tetraphenylphosphonium ion electrode; *Rhodobacter capsulatus*

1. Introduction

During the last three decades, tellurium compounds have been extensively used in applied chemistry, metallurgy, and electronic industry [1,2]. Consequently, the water-soluble oxyanion tellurite (TeO_3^{2-}), can be found in high concentrations in land and waters near sites of waste discharge of industrial manufacturing processes [2]. It has been reported that K_2TeO_3 is extremely toxic to mammalian cells [3] and microorganisms [4] at concentrations as low as $1 \mu\text{g/ml}$ ($4 \mu\text{M}$). Notably, several phototrophic bacteria have been shown to be resistant to a variety of metal and transition metals, including K_2TeO_3 , at concentrations as high as 2.7 mg/ml [5,6]. Most of the tellurite-resistant bacteria convert tellurite to elemental tellurium (Te^0) which is accumulated intracellularly as black inclusions [5,2] although tellurite resistance without Te^0 accumulation was observed in a few species (see [5]).

The facultative phototrophs *Rhodobacter capsulatus* B100

and *Rhodobacter sphaeroides* 2.4.1 are highly resistant to tellurite (minimal inhibitory concentrations on agar media of 250 and $150 \mu\text{g/ml}$ K_2TeO_3 , respectively [7,8]) and light-grown cells accumulate crystallites of Te^0 [7,8]. Notably, plasma membrane vesicles isolated from *R. capsulatus* B100 cells grown in the presence of K_2TeO_3 were shown to catalyze both photosynthetic and respiratory electron transport activities [7]. The latter results demonstrate that light-grown cells of *R. capsulatus* B100 accumulate Te^0 with no drastic effects on bacterial cell viability. This conclusion, along with the scarcity of information on the mechanism of uptake by bacteria (see [2]), prompted us to investigate the way the oxyanion tellurite can enter into cells of *R. capsulatus*.

In this work we show that tellurite uptake occurs at concentrations up to $120\text{--}160 \mu\text{M}$ ($30\text{--}40 \mu\text{g/ml}$); at higher concentrations, K_2TeO_3 increases the light-driven membrane potential ($\Delta\psi$) affecting the ΔpH component of the electrochemical proton motive force ($\Delta\mu_{\text{H}^+}$). As the K^+/H^+ exchanger nigericin inhibits tellurite uptake, we conclude that this latter process is ΔpH dependent and non-electrogenic. We also show that tellurite uptake involves a transporter(s) totally inhibited by the sulfhydryl reagent *N*-ethylethylmaleimide (NEM).

2. Materials and methods

2.1. Bacterial strain

R. capsulatus B100 was grown anaerobically in the light (200 W cm^{-2}) at 30°C in RCV-malate minimal salt medium as described previously [9]. Cells were grown to early stationary phase ($A_{660} = 1.2$) and used within 6 h of harvesting.

2.2. Determination of potassium tellurite uptake

The quantitative determination of K_2TeO_3 was done with the reagent diethyldithiocarbamate (DDTC) as in [10]. In this assay, DDTC was incubated with tellurite concentrations in the range of $1\text{--}50 \mu\text{g/ml}$ at neutral pH to form a yellow solution. The absorbance was read at 340 nm . Assay conditions: t , 28°C ; cells equivalent to 1 mg/ml of proteins or $30 \mu\text{g/ml}$ of bacteriochlorophyll (BChl); light intensity, 200 W cm^{-2} . Anaerobic medium (Gly–Gly 50 mM or phosphate buffer 0.1 M , MgCl_2 2.5 mM , KCl 10 mM , pH 7.4) was flushed for 1 h with pure N_2 gas before use. Inhibitor concentrations: ad hoc experiments (not shown) were performed to determine the optimal concentrations of the different inhibitors used in this study (see also [11–13]).

2.3. Electrode measurements of tetraphenylphosphonium (TPP^+) accumulation ($\Delta\psi$ determination)

A polyvinylchloride membrane, selectively permeable to TPP^+ , was constructed as described in [11]. An internal cell volume of $102 \mu\text{l}$ of BChl was assumed, according to Kell et al. [14].

2.4. Protein and BChl determinations

Protein content of the samples was determined by the method of

*Corresponding author. Fax: (39)-051-242576.

E-mail address: davide.zannoni@unibo.it (D. Zannoni).

Abbreviations: BChl, bacteriochlorophyll; DCCD, dicyclohexyl carbodiimide; DDTC, diethyldithiocarbamate; $\Delta\mu_{\text{H}^+}$, electrochemical proton gradient; DIDS, 4,4-diisothiocyanatostilbene-2,2-disulfonate; FCCP, carbonyl cyanide-*p*-trifluoromethoxyphenyl-hydrazone; NEM, *N*-ethylethylmaleimide; TPP^+ , tetraphenylphosphonium ion

Lowry [15] using bovine serum albumin (BSA) as a standard. The content of BChl was measured spectrophotometrically in acetone/methanol (7:2 v/v) extracts using an ϵ_{775} of $75 \text{ mM}^{-1} \text{ cm}^{-1}$ [16].

2.5. Polarographic measurements

Respiratory activities in intact cells were determined with a Clark-type oxygen electrode YSI 53 (Yellow Springs Instruments Inc., Yellow Springs, OH, USA) as detailed elsewhere [9].

3. Results

3.1. Tellurite uptake by light-grown cells of *R. capsulatus*

We have recently reported that light-grown cultures of *R. capsulatus* are resistant to K_2TeO_3 concentrations up to $50 \text{ }\mu\text{g/ml}$ [7]. Growth in the presence of K_2TeO_3 caused the bacteria to turn black, indicative of the presence of Te^0 , and we have also shown that Te^0 crystals are accumulated inside the cells where tellurite reduction occurs [7]. Here (Fig. 1a) the light-driven tellurite uptake kinetics by *R. capsulatus* cells as a function of variable tellurite concentrations (up to $25 \text{ }\mu\text{g/ml}$) are shown. Considering the dissociation constants of tellurous acid (3×10^{-3} and 2×10^{-8} for k_1 and k_2 , respectively), tellurite at neutral pH should be mainly present in the form of HTeO_3^- and TeO_3^{2-} with no Te^{4+} present due to its instability in water. Based on this, the Lineweaver–Burk plot shown in Fig. 1a (inset) is likely to indicate the apparent K_m of the $\text{HTeO}_3^-/\text{TeO}_3^{2-}$ uptake process (K_m of $20 \text{ }\mu\text{g/ml}$ or $80 \text{ }\mu\text{M}$). Under our assay conditions the tellurite uptake rate was $1.6 \pm 0.1 \text{ }\mu\text{g/min}$ per mg of protein, a value which is several times higher than the rate reported for *Escherichia coli* cells harboring tellurite-resistant determinants [15]. Notably, no uptake was seen at $\text{K}_2\text{TeO}_3 \geq 50 \text{ }\mu\text{g/ml}$ (200 nmol/mg of protein) (not shown). Fig. 1b shows that after 5 min of illumination, valinomycin, a mobile carrier catalyzing the electrical uniport of K^+ and dissipating the $\Delta\psi$, poorly affects the tellurite uptake (25% inhibition). Conversely, a strong inhibition is seen in the presence of the electroneutral exchanger nigericin (one K^+ for one H^+) (90% inhibition) and/or the proton translocator carbonyl cyanide-*p*-trifluoromethoxyphenyl-hydrazine (FCCP) (95% inhibition). These results indicate that tellurite is mainly taken inside the cells by a ΔpH -dependent process. Table 1 summarizes the rates of tellurite uptake in the presence of ionophores and of several reagents of sulfhydryl (NEM and mersalyl), carboxyl (dicyclohexyl carbodiimide, DCCD) and nucleophilic (4,4-diisothiocyanatostilbene-2,2-disulfonate, DIDS) groups. The mitochondrial phosphate carrier inhibitor NEM was greatly affected the tellurite uptake (95%) whereas another sulfhydryl reagent, mersalyl and the well-known inhibitor of the anion exchanger in mammalian

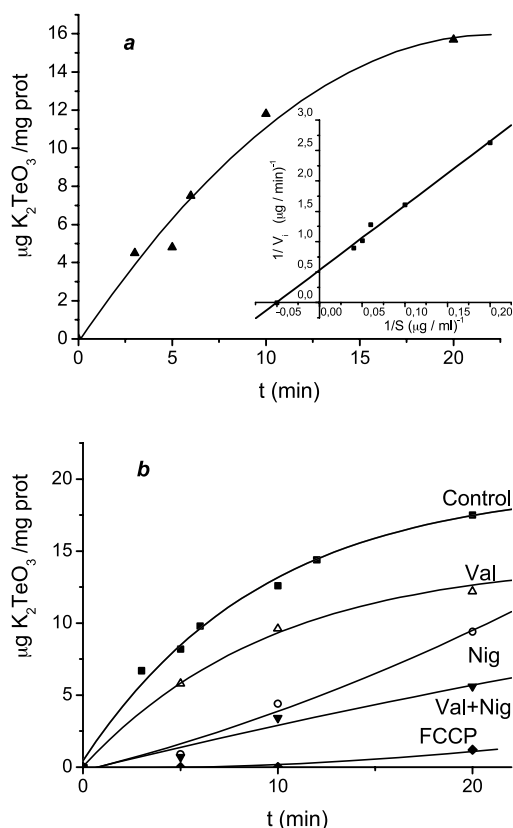


Fig. 1. a: Light-dependent tellurite uptake kinetic by *R. capsulatus* cells. Inset: Lineweaver–Burk plot. Assay conditions: $t = 28^\circ\text{C}$, anaerobic Gly–Gly buffer (50 mM), MgCl_2 2.5 mM, KCl 10 mM, pH 7.4. b: Tellurite uptake as a function of various ionophores: Val, valinomycin (4 μM); Nig, nigericin (3 μM); FCCP (5 μM). Assay conditions as in a except for the anaerobic buffer (phosphate 0.1 M). The apparent K_m was approximately 1 mM tellurite (not shown). See Section 2.2 for further details.

cells DIDS [12] were less effective (40–50% inhibition). DCCD, an inhibitor of the ATPase proton channel [13] but also of the redox complex QH_2 :cytochrome *c* oxidoreductase [18] involved in $\Delta\mu_{\text{H}^+}$ generation, had a strong inhibitory effect (70%) on tellurite uptake.

3.2. Tellurite effect on membrane potential generated by cells of *R. capsulatus*

To further verify the kinetic data of Fig. 1 indicating that tellurite uptake is not driven by the $\Delta\psi$, the distribution of the lipophilic cation TPP^+ was employed to estimate the $\Delta\psi$ in

Table 1
Effect of various ionophores and inhibitors on tellurite uptake

	TeO_3^{2-} uptake activity ^a ($\mu\text{g/min}$ per mg of protein)	% of activity
Control	1.60 ± 0.08	100 ± 5
+FCCP (5 μM)	0.08 ± 0.03	5 ± 2
+Nigericin (3 μM)	0.16 ± 0.08	10 ± 5
+Valinomycin (4 μM)	1.20 ± 0.08	75 ± 5
+DIDS (50 μM)	0.80 ± 0.32	50 ± 20^b
+DCCD (100 μM)	0.50 ± 0.08	30 ± 5
+NEM (0.5 mM)	0.08 ± 0.03	5 ± 5
+Mersalyl (0.2 mM)	0.95 ± 0.08	60 ± 5

^aThe uptake rates were determined on a 5 min time period of continuous illumination (see Section 2 and legend of Fig. 1b for details).

^bThe high value of statistical error is due to the optical interference at 340 nm between DIDS and the TeO_3^{2-} determination reagent (DDTC) (see also Section 2).

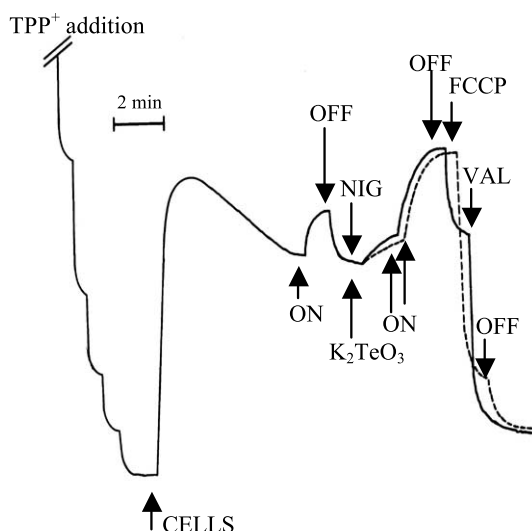


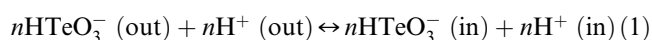
Fig. 2. Light-induced and oxygen-dependent uptake of TPP^+ ions by *R. capsulatus* cells. Following the calibration addition of TPP^+ (1 μM final concentration), cells (45 $\mu\text{g}/\text{ml}$ of BChl) were added to 2 ml of air-saturated medium (TES buffer, 50 mM pH 7.5, KCl 10 mM, at 28°C). Symbols: ON, light-on; OFF, light-off. Abbreviations as in Fig. 1. See Section 2.3 and [11] for details.

cells of *R. capsulatus* under the same conditions used for the uptake kinetic measurements. Fig. 2 shows (continuous trace) that following the addition of *R. capsulatus* cells, a rapid upward deflection of the trace, indicative of TPP^+ uptake (development of a negative potential inside the cells), can be seen. Apparently, the initial level of $\Delta\psi$ cannot be maintained by respiration possibly due to the rate-limiting oxygen diffusion through the external membrane/wall structure of *R. capsulatus*. Under steady-state respiratory conditions (reached after approximately 5 min) the estimated $\Delta\psi$ was 104 ± 2 mV, while under both respiration and continuous illumination (light on), the $\Delta\psi$ went up to 137 ± 2 mV. Respiratory measurements, performed in parallel with TPP^+ uptake determinations, indicated that the endogenous respiration by cells of *R. capsulatus* is severely inhibited by light (80%, not shown). This suggests, in line with early reports [11], that the light-generated $\Delta\psi$ exhibits such a control over respiration that the latter process poorly contributes to $\Delta\psi$ formation when cells are subjected to continuous illumination. The zero $\Delta\psi$ was therefore defined by the TPP^+ uptake levels seen in the dark after the final addition of FCCP (1.5 μM). Fig. 2 also shows that addition of the K^+/H^+ exchanger nigericin (3 μM) increased the $\Delta\psi$ formed by both respiration and photosynthesis ($\Delta\psi = 147 \pm 2$ mV), through dissipation of the ΔpH component of the $\Delta\mu_{\text{H}^+}$. As expected, addition of valinomycin (4 μM) totally collapsed the TPP^+ uptake. Interestingly, also the addition of tellurite (50 $\mu\text{g}/\text{ml}$, i.e. 200 μM) increased the light-generated $\Delta\psi$ (interrupted trace), thus mimicking the effect of nigericin (see Section 4 for further discussion on this point).

4. Discussion and conclusions

This study examines the light-induced uptake of the metalloid oxyanion tellurite by cells of the facultative phototroph *R. capsulatus*. Results clearly indicate that the uptake is an energy-dependent process strongly repressed by the uncoupler

FCCP and/or the K^+/H^+ exchanger nigericin but much less by the K^+ ionophore valinomycin. This suggests that the movement of tellurite from the external medium into cell cytoplasm, where tellurite reduction occurs [7], is an electroneutral mechanism driven mainly by the ΔpH . Tellurite uptake was repressed by the sulfhydryl reagent, phosphate carrier inhibitor NEM and it was partially sensitive to the oxyanion transporter inhibitor DIDS. Further, the apparent K_m for tellurite uptake varied from 80 μM to approximately 1 mM in the presence of 0.1 M phosphate (not shown). These observations, taken together, bring us to the conclusion that the uptake of tellurite by light-grown *R. capsulatus* is likely to be mediated by the phosphate transport system. In line with this it has previously been shown that the transport of phosphate (Pi) in *E. coli* is inhibited by tellurite [19]. Further, mutants of *E. coli* defective in Pi transport were resistant to levels of tellurite considerably higher (10–40 times) than those observed in most wild-type strains. Susceptibility to tellurite could be restored by a plasmid carrying the *phoB* region, which is involved in Pi regulation. Apparently, these early reports strongly support our proposal that *R. capsulatus* takes up tellurite by a Pi transporter, possibly of the PiT family [20]. In this respect, at neutral pH, we suggest the following generalized transport reaction:



During the past decade, several determinants of tellurite resistance (Te^R) have been cloned and sequenced [8,17,21–23]. The chromosome of *E. coli* K12 contains two genes, named *tehA* and *tehB*, encoding for two polypeptides TehA and TehB. The TehA protein is highly hydrophobic [24] and it shows homology with the multidrug transporters of the SMR family. Although the connection between the phenotypes of cationic multidrug resistance and Te^R along with the biochemical nature of Te^R are not yet understood, the mechanisms of reduced uptake and/or efflux of TeO_3^{2-} have been ruled out [17]. As the latter conclusion is referred to *E. coli*, the mechanism of tellurite uptake here described for *R. capsulatus* can hardly be linked to the membrane polypeptides previously shown to be associated with Te^R in various bacterial species [2].

In conclusion, we believe that the present study not only contributes to a better understanding of how tellurite is taken up by light-grown cells of *R. capsulatus* but it also gives further insights into the multiple metabolic features of facultative phototrophs.

Acknowledgements: We wish to thank Mario Mancon for technical assistance on the use of the TPP^+ ion exchange electrode and MURST of Italy for financial support (PRIN2001).

References

- [1] Klevay, L.M. (1976) *Pharmacol. Ther.* 1, 223–229.
- [2] Taylor, D.E. (1999) *Trends Microbiol.* 7, 111–115.
- [3] Wagner, M., Toews, A.D. and Morell, P. (1995) *J. Neurochem.* 64, 2169–2176.
- [4] Turner, R.J. (2001) *Rec. Res. Dev. Microbiol.* 5, 69–77.
- [5] Yurkov, V.V. and Beatty, J.T. (1998) *Microbiol. Mol. Biol. Rev.* 62, 695–724.
- [6] Moore, M.D. and Kaplan, S. (1992) *J. Bacteriol.* 174, 1505–1514.
- [7] Borsetti, F., Borghese, R., Francia, F., Randi, M.R., Fedi, S. and Zannoni, D. (2003) *Protoplasma* 221, 153–161.

- [8] O'Gara, J.P., Gomelsky, M. and Kapan, S. (1997) *Appl. Environ. Microbiol.* 63, 4713–4720.
- [9] Zannoni, D., Jasper, P. and Marrs, B.L. (1978) *Arch. Biochem. Biophys.* 191, 625–631.
- [10] Turner, R.J., Weiner, J.H. and Taylor, D.E. (1992) *Anal. Biochem.* 204, 292–295.
- [11] Rugolo, M. and Zannoni, D. (1983) *Biochem. Biophys. Res. Commun.* 113, 155–162.
- [12] Cabantchik, Z.I., Knauf, P.A. and Tothstein, A. (1978) *Biochim. Biophys. Acta* 515, 239–302.
- [13] Downie, J.A., Gibson, F. and Cox, G.B. (1979) *Annu. Rev. Biochem.* 48, 103–132.
- [14] Kell, D.B., Ferguson, S.J. and John, P. (1978) *Biochim. Biophys. Acta* 502, 11–126.
- [15] Lowry, O.H., Rosebrough, N.J., Farr, A.L. and Eandall, R.J. (1951) *J. Biol. Chem.* 193, 265–275.
- [16] Clayton, R.K. (1963) *Biochim. Biophys. Acta* 75, 312–323.
- [17] Turner, R.J., Weiner, J.H. and Taylor, D.E. (1995) *Can. J. Microbiol.* 41, 92–98.
- [18] Schinkarev, V.P., Ugulava, N.B., Crofts, A.R. and Wraight, C.A. (2000) *Biochemistry* 39, 16206–16212.
- [19] Tomas, J.M. and Kay, W.W. (1986) *Antimicrob. Agents Chemother.* 30, 77–101.
- [20] Harris, R.M., Webb, D.C., Howitt, S.M. and Cox, G.B. (2001) *J. Bacteriol.* 183, 5008–5014.
- [21] Turner, R.J., Weiner, J.H. and Taylor, D.E. (1994) *Microbiology* 140, 1319–1326.
- [22] Whelan, K.T., Sherburne, R.K. and Taylor, D.E. (1997) *J. Bacteriol.* 178, 63–71.
- [23] Cournoyer, B., Watanabe, S. and Vivian, A. (1998) *Biochim. Biophys. Acta* 1397, 161–168.
- [24] Turner, R.J., Taylor, D.E. and Weiner, J.H. (1997) *Antimicrob. Agents Chemother.* 41, 440–444.